

SOPHIA: A mineralogical simulant for phyllosilicate terrains at the *Rosalind Franklin* landing site, Oxia Planum, Mars.

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Abstract

The phyllosilicate-bearing martian plain, Oxia Planum, is the proposed landing site for the *Rosalind Franklin* rover mission, scheduled to launch in 2028. *Rosalind Franklin* which will search for signs of past or present life on Mars. Terrestrial analogue sites and simulants can be used to test instruments analogous to those on *Rosalind Franklin*, however no simulant for Oxia Planum currently exists. In anticipation of *this mission*, a simulant - SOPHIA (Simulant for Oxia Planum: Hydrated, Igneous, and Amorphous) - representative of the local mineralogy at Oxia Planum has been developed for biosignature and mineralogy experiments, which will assist in interpreting data returned by the rover. The simulant is derived from orbital observations of Oxia Planum and its catchment area. As no *in situ* data is available for Oxia Planum, mineralogy from other comparable sites on Mars was used to design the simulant including orbital data from Arabia Terra and Mawrth Vallis and *in situ* data collected from Gale crater. The mineralogy, chemistry and physical properties of the simulant were characterised

30 using standard laboratory techniques (SEM-EDS, XRF, XRD). Techniques analogous to rover
31 instruments (Raman spectroscopy, Near-IR spectroscopy analogous to the Raman laser spectrometer
32 and ISEM and MicrOmega instruments) were also used. The simulant is rich in Fe/Mg phyllosilicates
33 with additional primary igneous and other alteration minerals and is an appropriate spectral and
34 mineralogical analogue for Oxia Planum.

35 **Keywords**

36 Mineralogy, Mars, Surface, Regolith, Astrobiology.

37 **1 Introduction**

38 Oxia Planum is the landing site for the European Space Agency's Mars rover, *Rosalind Franklin*. The
39 rover will search for evidence of past or present life and characterise the geochemical environment
40 on the surface and in the near subsurface (Vago et al. 2017). *Rosalind Franklin* is designed to identify
41 both physical biosignatures, such as microfossils (García-Ruiz et al. 2003; Schopf et al. 2012) and
42 microbially-induced sedimentary structures (Noffke et al. 2001, 2013) and chemical biosignatures,
43 including biominerals, isotopes and organic molecules (biomarkers), that are indicative of processes
44 unique to life (Bazylinski and Moskowitz 1997; Beard et al. 1999; Banfield et al. 2001; Simoneit 2004;
45 Georgiou and Deamer 2014; Moreras-Marti et al. 2022). To identify biomarkers, the rover will use a
46 specifically-selected suite of instruments including the Mars Organic Molecule Analyser (MOMA)
47 which includes Gas Chromatography-Mass and Laser Desorption-Mass Spectrometry (GCMS and
48 LDMS; Goesmann et al. 2017) and the Raman Laser Spectrometer (RLS; Rull et al. 2017). To study the
49 mineralogy of the site, RLS will be used in conjunction with the Infrared spectrometer for ExoMars
50 (ISEM; Korablev et al. 2017), a visible near-infrared hyperspectral microscope (MicrOmega; Bibring et
51 al. 2017)), the Mars Multispectral Imager for Subsurface Studies (Ma_MISS; De Sanctis et al. 2017,
52 2022), multispectral Panoramic cameras with geologically specific filters (PanCam; Cousins et al. 2012;
53 Harris et al. 2015; Coates et al. 2017) and the Close-UP Imager (CLUPI; Josset et al. 2017).

54 Despite the application of this instrument suite, biomarkers may remain undetected or unrecognised
55 because they can be degraded or modified over geological time by processes such as UV and cosmic
56 radiation (Kminek and Bada 2006; Dartnell et al. 2007, 2012), burial, diagenesis (Tan and Sephton
57 2020) and impacts (Parnell et al. 2005; Bowden et al. 2009; Montgomery et al. 2016). For example,
58 variance in the shock impedance of minerals has been shown to create microscale pressure and
59 temperature environments that may degrade biomarkers to variable degrees; degradation of amino
60 acids has been shown to be higher in a haematite host mineralogy than in olivine (Furukawa et al.
61 2018). Local mineralogy can also control the detectability of organic species. For example, highly

62 absorbent or oxidising minerals can make the extraction and detection of biomarkers difficult (Röling
63 *et al.* 2015) and the presence of perchlorates on the martian surface (Hecht *et al.* 2009; Glavin *et al.*
64 2013; Clark *et al.* 2021) has been proposed to modify organic species during the py-GC-MS
65 experiments on the Curiosity rover (Millan *et al.* 2020); perchlorates can become highly oxidizing and
66 may destroy or transform biomarkers into chlorinated species (Steininger *et al.* 2012; He *et al.* 2020).
67 Thus, the mineral matrix which hosts biomarkers is important for understanding their preservation
68 and detectability.

69 Mineralogy can also play a role in the preservation and longevity of biomarkers. Importantly,
70 phyllosilicate minerals have been shown to adsorb organic molecules into the layered structure
71 (Hedges 1977), protecting them from degradation by radiation and oxidation (Poch *et al.* 2015; dos
72 Santos *et al.* 2016; Ertem *et al.* 2017). Phyllosilicates have also been shown to preserve amino acids
73 from degradation by UV radiation (dos Santos *et al.* 2016). Phyllosilicates have been detected from
74 orbit at Oxia Planum by the Observatoire pour la Minéralogie, l'Eau, les Glaces et l'Activité (OMEGA;
75 Bibring *et al.* 2005) and Compact Reconnaissance Imaging Spectrometer for Mars (CRISM; Murchie *et al.*,
76 2007) (i.e., Carter *et al.*, 2016; Mandon *et al.*, 2016; Brossier *et al.*, 2022), providing evidence of
77 aqueous activity within the region that could indicate it was once habitable (Carter *et al.* 2016; Vago
78 *et al.* 2017). This terrain is the primary target of the *Rosalind Franklin* mission because of this organic
79 preservation potential (Parnell *et al.* 2007; Fraser *et al.* 2011; Poch *et al.* 2015; Broz 2020).

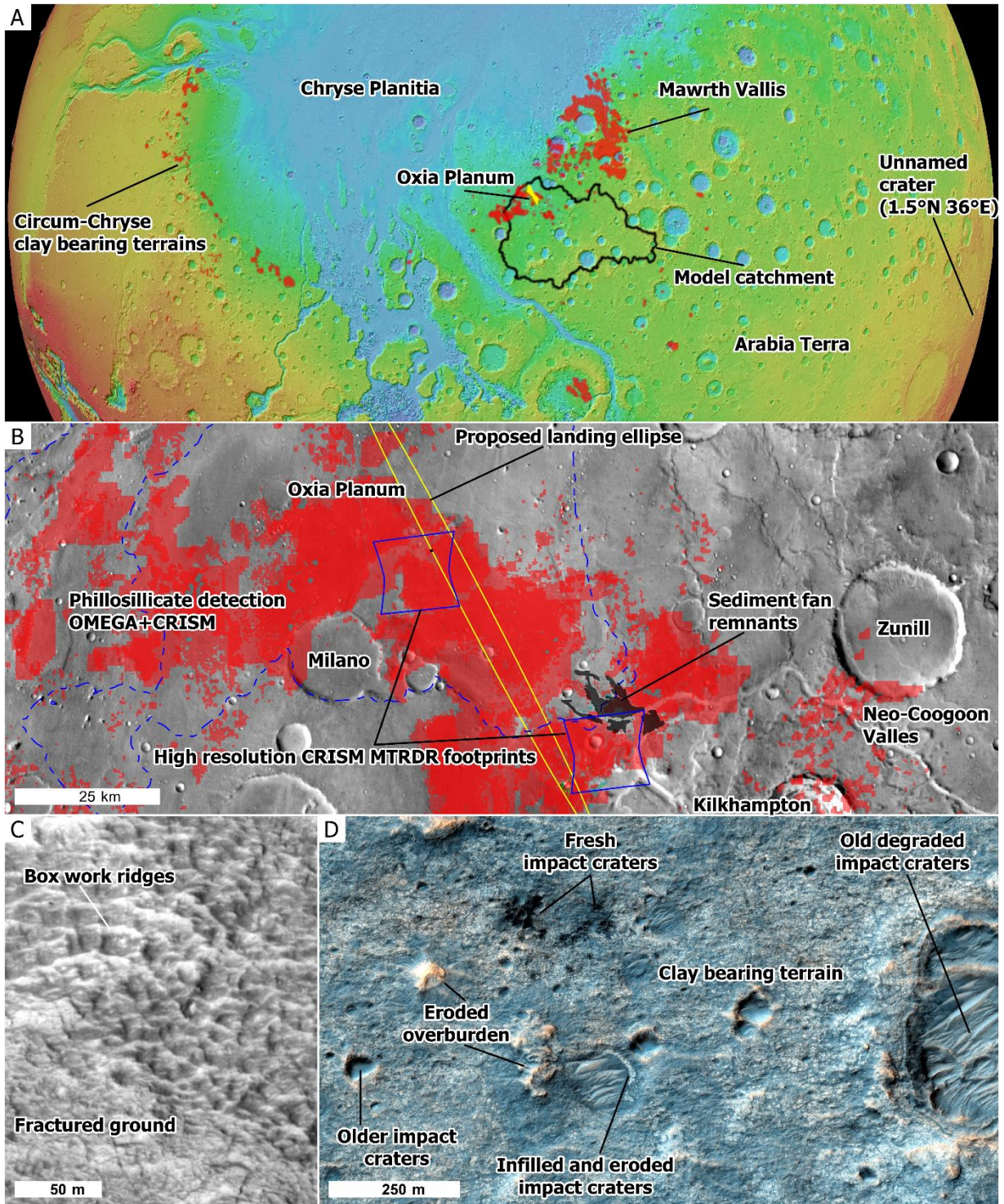
80 In anticipation of landing at Oxia Planum, and in the absence of direct samples or analyses of this
81 region of Mars, terrestrial analogues and simulants are essential for experimentation and simulation
82 in preparation for interpretation of returned data. The use of an accurate simulant will aid the
83 interpretation of data returned from the rover to identify the mineralogy at Oxia Planum. The
84 development of a simulant for Oxia Planum could assist in understanding the preservation and
85 alteration potential of biosignatures at Oxia Planum and identify appropriate biosignature targets for
86 the site. In this paper we present a new simulant, SOPHIA (Simulant for Oxia Planum: Hydrated,
87 Igneous, Amorphous), which represents the mineralogy of Oxia Planum's phyllosilicate-bearing
88 terrain.

89 1.1 Geological context – Oxia Planum

90 Oxia Planum is a phyllosilicate-bearing basin (Figure 1b) located on the edge of Mars' crustal
91 dichotomy where the southern highlands of western Arabia Terra meet the northern lowlands, at the
92 southern margin of Chryse Planitia (Figure 1a). Oxia Planum has undergone several periods of fluvial
93 activity associated with Coogoon Vallis and has fluvial systems (Hynek *et al.* 2010; Molina *et al.* 2017;
94 Quantin-Nataf *et al.* 2021; Fawdon *et al.* 2022). The phyllosilicate deposits in the basin, detected by

95 CRISM and OMEGA (Carter *et al.* 2016; Mandon *et al.* 2021; Brossier *et al.* 2022), occur where basaltic
96 minerals, emplaced during the Noachian (4.1 – 3.7 Gya), were altered in subaqueous or subaerial
97 environments either *in situ* or during transport to the basin *via* the surrounding channel network
98 (Quantin-Nataf *et al.* 2021). However, the exact formation mechanism is difficult to deduce from
99 orbital data alone (Carter *et al.* 2016; Mandon *et al.* 2021; Quantin-Nataf *et al.* 2021; Brossier *et al.*
100 2022). In recent mapping efforts by the Rover Science Operations Working Group (RSOWG), these
101 terrains have been described as the upper and lower bedrock groups (Fawdon *et al.* 2023). This
102 phyllosilicate-bearing unit forms part of a widespread network of phyllosilicate deposits that extend to
103 the Chryse Planitia region. While the connection between these deposits is unclear but may represent
104 part of an ancient shoreline of an ocean extending into the northern lowlands if one existed (Figure
105 1a and c; Parker *et al.* 1989, 1993; Head *et al.* 1998; Clifford and Parker 2001; Ivanov and Head 2001;
106 Dickeson and Davis 2020).

107 Aqueous episodes in Oxia Planum that post-date the Noachian phyllosilicates (Fawdon *et al.* 2022) are
108 associated with a sedimentary fan deposit containing hydrated silica (Pan *et al.* 2021). The
109 sedimentary fan is thought to have formed during reactivation of the channel systems within Coogoon
110 Vallis (Fawdon *et al.* 2021; Quantin-Nataf *et al.* 2021). An overlying dark deposit caps the
111 phyllosilicates and fan deposits and may have aided their preservation and that of possible
112 biosignatures within them (Quantin-Nataf *et al.* 2021).



113

114 **Figure 1:** Context. (A) global context of Oxia Planum, and the Circum-Chryse phyllosilicate-bearing terrains (red;
 115 (Carter et al. 2013, 2015) that SOPHIA represents. Oxia Planum is located on the border of Chryse Planitia, with
 116 the Mawrth Vallis region to the east. The un-named crater from Lai et al. (2019) is shown to the southeast. (B)
 117 The local context of phyllosilicate-bearing terrains as identified by the OMEGA and CRISM (mapping mode)
 118 instruments (RED) in the Oxia basin (outlined by the -3000m contour; dashed blue). Also shown are high
 119 resolution CRISM Map-Projected Targeted Reduced Data Records (MTRDR) footprints, the sediment fans
 120 associated with Neo-Coogoon Vallis and a -3000m contour line delineating 'The Oxia Basin' (Fawdon et al. 2022)
 121 (C) An example of a network of upstanding ridges interpreted to be mineralised veins indicating ground water
 122 activity in Oxia Planum (HiRISE (High Resolution Imaging Science Experiment) image ESP_037558_1985). (D) The
 123 variety of small (<500 m) scale impact structures that have affected the phyllosilicate-bearing terrains in Oxia

124 *Planum, which includes contemporary impact craters retaining dark haloes, older partially eroded impact craters,*
125 *and much older impact structure that have heavily eroded but also show evidence for having been infilled with*
126 *some now lithified material (HiRISE image ESP_062402_1985_RGB).*

127 **1.2 Previous Mars simulants**

128 In the absence of returned samples from Mars, simulants are used to represent its physical, chemical
129 and/or mineralogical properties. Simulants are used in experiments or mission preparation and testing
130 prior to launch and for interpretation of mission data. Martian simulants have been developed to
131 represent global martian mineralogy mainly focusing on the planet's igneous mineralogy: JSC Mars-1
132 was developed based on spectroscopic data collected by Phobos 2 (Allen *et al.* 1998) and is mostly
133 composed of terrestrial volcanic ash (Allen *et al.* 1998) and the Mojave Mars Simulant (MMS), a
134 geotechnical simulant, was developed using data from the Viking landers (Banin *et al.* 1992),
135 Pathfinder (Moore *et al.* 1999; Foley *et al.* 2003) and the Mars Exploration Rovers (e.g. Klingelhöfer *et*
136 *al.* 2004; Morris *et al.* 2006a, 2006b; Yen *et al.* 2005), and is composed of mechanically crushed
137 boulders (Peters *et al.* 2008). MSG-1 was designed to represent the Rocknest soil at Gale crater, which
138 is accepted to be representative of a typical basaltic mineral composition on Mars (Bish *et al.*, 2013),
139 containing amorphous material in addition to a crystalline component (Cannon *et al.* 2019).

140 Although these simulants are globally relevant, there are significant regional differences in geology on
141 Mars, and areas where geological processes (including aqueous alteration) have resulted in diverse
142 mineral assemblages (Bandfield 2002; Mustard 2005; Bibring *et al.* 2006; Wang *et al.* 2006; Ehlmann
143 *et al.* 2008; Carter *et al.* 2010, 2013; Ody *et al.* 2013; Ehlmann and Edwards 2014; Salvatore *et al.* 2018;
144 Osterloo *et al.*, 2008). These are not represented in 'global' Mars simulants meaning those simulants
145 are insufficient for studies investigating specific local environmental conditions or for addressing
146 questions pertaining to the habitability and preservation potential of these terrains.

147 More recent simulants represent local mineralogy; for example, the S-MRS (Böttger *et al.* 2012)
148 represents a sulphate-rich terrain where sulphates, as detected from orbit and *in situ* (Klingelhöfer *et*
149 *al.* 2004; Bibring *et al.* 2005; Wang *et al.* 2006), formed during the Hesperian when water evaporated
150 from the martian surface. The OUXX-1 simulants (Ramkissoon *et al.* 2019) were based on four distinct
151 compositions: 1) an early Mars basaltic terrain (based on a basaltic shergottite meteorite composition
152 (Bridges and Warren 2006); 2) a sulphur-rich simulant (based on Paso Robles sulphate-rich samples
153 analysed by the *Spirit* rover at Columbia Hills (Gellert *et al.* 2006; Morris *et al.* 2006b); 3) a haematite-
154 rich simulant (based on Hematite Slope, a haematite-rich sample at Meridiani Planum analysed by the
155 *Opportunity* rover); 4) a contemporary Mars simulant (based on Rocknest in Gale crater, analysed by
156 the *Curiosity* rover). The Y-Mars simulant (Stevens *et al.* 2018) was developed to represent the
157 composition of the mudstone in the Sheepbed member at Yellowknife Bay, Gale crater, using data
158 from CheMin on board the *Curiosity* rover. The P-MRS simulant (Böttger *et al.* 2012) was developed

159 to represent the mineralogy of a phyllosilicate-bearing terrain on Mars in which igneous rocks had
160 been altered to phyllosilicate by pH neutral hydrous fluids in a CO₂ rich atmosphere.

161 Despite these more specific simulants, none have been developed to represent the geological context
162 and mineralogy at Oxia Planum. The P-MRS simulant (Böttger *et al.* 2012) was based on data from
163 across Mars and not an individual region. In addition, it contains 45 % montmorillonite, a phyllosilicate
164 associated with extensive weathering that is not observed to be abundant at Oxia Planum from orbit.
165 Therefore, the development of a simulant specific to Oxia Planum is required so that its precise
166 environment and local variations, specifically phyllosilicate-bearing terrain, can be accommodated.
167 This will help to inform the exploration and data analysis by the rover and studies of this site's
168 astrobiological potential.

169 2 Simulant design

170 2.1 Oxia Planum and comparison sites

171 To initially evaluate the mineralogy needed for SOPHIA, spectral data for Oxia Planum from orbiting
172 spacecraft were considered. The phyllosilicate-bearing terrain at Oxia Planum (Figure 1b) is dominated
173 by Fe/Mg phyllosilicates, as indicated by absorption bands at 1.41, 1.92, 2.30 μm measured by CRISM
174 and OMEGA (Clark *et al.* 1990; Carter *et al.* 2016; Mandon *et al.* 2021; Brossier *et al.* 2022). These
175 absorption bands match best to vermiculite or saponite; vermiculite has absorption bands at 1.42 μm,
176 2.30-2.31 μm and 2.39 μm (Clark *et al.* 1990; Carter *et al.* 2013; Swayze *et al.* 2018) while Fe-Saponite
177 has absorption bands at 1.40-1.42 μm, 2.30-2.32 μm and 2.4 μm (Treiman *et al.* 2014). Given the
178 similarities in their spectra, it is difficult to distinguish which may be present at Oxia Planum (Mandon
179 *et al.* 2021; Brossier *et al.* 2022).

180 A further complexity in identifying Oxia Planum's phyllosilicate mineralogy comes from the
181 identification of two spectrally diverse sub-terrains, potentially resulting from the mixing of
182 phyllosilicates with other minerals. In the east, CRISM suggests the phyllosilicate terrain is enriched
183 with olivine, since it is dominated by Fe²⁺-rich material (Mandon *et al.* 2021; Parkes Bowen *et al.* 2022).
184 In the middle and west of the landing site, CRISM (Mandon *et al.* 2021; Parkes Bowen *et al.* 2022) has
185 suggested the phyllosilicate may be Fe³⁺-rich, mixed with a Fe-smectite and/or serpentine, chlorite,
186 and/or a carbonate (possibly siderite) identified by a feature at 2.5 μm in the CRISM spectra. As the
187 material to the west and middle of Oxia Planum dominates the landing ellipse for the rover, these
188 terrains will be represented by the SOPHIA simulant. However, as reflectance spectroscopy is highly
189 sensitive to phyllosilicate minerals, the material at Oxia Planum may include a range of compositions
190 from phyllosilicate rich to phyllosilicate poor. To represent a phyllosilicate rich scenario, Fe/Mg

191 phyllosilicates included as a major component in addition to siderite, Fe-smectite and/or serpentine
192 and chlorite.

193 However, some minerals cannot be detected within the spectral range of orbiting instruments. For
194 example, CRISM cannot detect non-Fe-bearing felsic minerals as they do not absorb in the Near
195 infrared (Milam *et al.* 2010). Additionally, orbital instruments have a lower spatial resolution (e.g.,
196 CRISM 15-19 m/pixel) than rover instruments, meaning some minerals present may not be visible from
197 orbit. To counteract these issues, the Oxia Planum mineralogy has been evaluated in the context of
198 data from the surrounding area. Specifically, CRISM data from the modelled fluvial catchment area of
199 the Oxia Basin (Brossier *et al.* 2022; Turner *et al.*, *Submitted*) (Figure 1b) has been incorporated to
200 specify the mineralogy for the simulant, assuming this material would have been transported into the
201 basin. To better inform decisions regarding the anhydrous (bedrock) mineralogy at Oxia Planum, data
202 from OMEGA and the Thermal Emission Spectrometer (TES) onboard Mars Global Surveyor, taken
203 from the floor of an unnamed crater where bedrock is exposed through an observational window in
204 Arabia Terra (Lai *et al.*, 2019) and olivine, pyroxene and Fe-glass have been detected, were used
205 (Figure 1a).

206 The minerals suggested to be present at Oxia Planum were compared with other phyllosilicate-bearing
207 sites on the martian surface. Mawrth Vallis was chosen as a comparison site because it contains part
208 of a larger phyllosilicate deposit in the circum Chryse Planitia region, where mineralogy has been
209 extensively studied from orbit (Bishop *et al.* 2008, 2020; Michalski *et al.* 2010; Bishop and Rampe 2016;
210 Lowe *et al.* 2020). At Mawrth Vallis, absorption bands at 1.91 and 2.29-2.31 μm were associated with
211 Fe/Mg phyllosilicates, possibly Fe-smectites and/or vermiculite (Bishop *et al.* 2008; McKeown *et al.*
212 2009). These Fe/Mg phyllosilicates have been suggested to be spectrally comparable to those at Oxia
213 Planum and in the wider Chryse Planitia region (Figure 1) (McKeown *et al.* 2009; Noe Dobrea *et al.*
214 2010, 2011; Loizeau *et al.* 2015; Baker 2017; Bishop *et al.* 2020; Lowe *et al.* 2020; Poulet *et al.* 2020).
215 At Mawrth Vallis, Fe/Mg phyllosilicates are overlain by Al-phyllosilicates (montmorillonite and/or
216 beidellite). These overlying phyllosilicates are suggested to have formed through the pedogenesis of
217 the aqueously altered Fe/Mg phyllosilicates (Bishop *et al.* 2008; McKeown *et al.* 2009; Bishop and
218 Rampe 2016; Liu *et al.* 2021). Since Al-phyllosilicates are not present in the areas proposed as the
219 landing site (Carter *et al.* 2016; Mandon *et al.* 2021; Brossier *et al.* 2022), these are not considered
220 and only the Fe/Mg phyllosilicates are used to determine the mineralogy relevant to the simulant.

221 Yellowknife Bay, Gale crater has also been chosen as a comparison site to Oxia Planum, in particular
222 the Sheepbed member, a mudstone with fine grained phyllosilicates. This mudstone has been
223 analysed (as drill fines) using the XRD on the Curiosity rover. The John Klein (JK) and Cumberland (CB)

224 drill holes contain trioctahedral Fe/Mg phyllosilicates (Vaniman *et al.* 2014). At Yellowknife Bay,
225 phyllosilicates are suggested to have formed in a closed-system, with circumneutral pH and minimal
226 oxidation (Vaniman *et al.* 2014; Bristow *et al.* 2018) in contrast to other environments at Gale crater,
227 e.g., Glen Torridon, which shows evidence of open-system, acidic and oxidizing environments
228 (Grotzinger 2014; Rampe *et al.* 2020). Similarities between the metre-scale fracturing at Yellowknife
229 Bay, observed by the MastCam instrument (Caswell and Milliken 2017), and the fracturing at Oxia
230 Planum, observed from orbit by HiRISE (Parkes Bowen *et al.* 2020; Apuzzo *et al.* 2022; Parkes Bowen
231 *et al.* 2022), also suggest similarities between these two locations, with the fracturing patterns
232 potentially resulting from similarities in grain size between the two sites or a common fracture
233 formation mechanism such as hydraulic fracturing (Parkes Bowen *et al.* 2020). The use of Yellowknife
234 Bay as a comparison site is limited by the large geographical distance between it and Oxia Planum. In
235 addition, the Fe/Mg phyllosilicates at Yellowknife Bay are much less extensive than those at Oxia
236 Planum and were not detected from orbit (Dehouck *et al.* 2017), meaning the extent of water-rock
237 interactions at Oxia Planum may be much less if phyllosilicates at both sites formed *in situ*. Despite
238 this, the JK and CB samples are used in the design of this simulant as they provide the most similar
239 phyllosilicates to those at Oxia Planum where subsurface *in situ* data is available.

240 This simulant is designed pre-emptive of rover interpretation of the site and, as such, the design
241 process used a holistic approach to understanding the mineralogy at Oxia Planum, which
242 encompassed its geological context, relationship to the wider Chryse Planitia and similarity to other
243 phyllosilicate terrains on Mars. The location specificity of the simulant means biosignature and
244 habitability studies can be directly relevant to the upcoming *Rosalind Franklin* rover. The simulant is
245 most representative of the lower bedrock groups at Oxia Planum where phyllosilicates have been
246 detected from orbit, and where olivine is less abundant.

247 2.2 Simulant Mineralogy

248 Using the data as described in the previous section, decisions regarding the specific mineralogy of the
249 simulant were as follows.

250 To determine the primary mineralogy of the simulant (the precursor mineralogy to the phyllosilicates
251 and other secondary minerals), data from an unnamed crater in Arabia Terra was used where igneous
252 bedrock is exposed. There, olivine and Ca-pyroxene have been detected (Lai *et al.*, 2019), consistent
253 with CRISM data for the valley walls of Coogoon Vallis in the Oxia Planum catchment area (Turner *et al.*,
254 *submitted*), and thus both minerals selected for inclusion in the SOPHIA simulant. Olivine is also
255 detected in the phyllosilicate -bearing unit at Oxia Planum; the width of the absorption band at Oxia
256 Planum is consistent with either fayalitic (Fe-rich) olivine or large grains of forsteritic (Mg-rich) olivine

257 (King and Ridley 1987; Mandon *et al.* 2021). As neither is more likely, forsteritic olivine was chosen for
258 the simulant as it was more easily obtained. K-feldspar was detected by *Curiosity* at JK and CB, but not
259 from orbit (Vaniman *et al.* 2014); however, the abundance of K-feldspar on Mars is thought to be
260 underrepresented in orbital data because of its limited reflectance in the near-IR making it
261 undetectable in CRISM spectra (Milam *et al.* 2010). For this reason, K-feldspar was also selected for
262 SOPHIA.

263 For the Fe/Mg phyllosilicate component, vermiculite and saponite were considered as they best
264 matched the dominant phyllosilicate mineralogy at Oxia Planum (Mandon *et al.* 2021; Brossier *et al.*
265 2022). Either of these minerals could have been chosen, but vermiculite was selected for the simulant
266 because it was more affordable. Spectral features at 2.5 μm suggest the dominant phyllosilicate
267 mineralogy is mixed with a second phyllosilicate mineral possibly serpentine or smectite, with
268 serpentine not easily assessed in the CRISM data (Leask *et al.* 2018). Serpentine was more easily
269 sourced so was included.

270 As the basaltic minerals at Oxia Planum discussed above are unlikely to be entirely weathered, an
271 intermediate mineral was required. Biotite and chlorite are phyllosilicates from which vermiculite can
272 form. Chlorite forms trioctahedral vermiculite in reducing conditions (Ross and Kodama 1973) and has
273 been detected in association with Fe/Mg phyllosilicates at Mawrth Vallis (Noe Dobrea *et al.* 2010;
274 Bishop *et al.* 2020) and was therefore favoured, however it could not be sourced within budget and
275 at the purity required and was replaced by biotite.

276 Other minerals commonly formed in association with Fe/Mg phyllosilicates were considered for the
277 simulant. Haematite forms through the oxidative aqueous alteration of basaltic material and is found
278 in low abundance in reducing circumneutral terrains on Mars, including at Yellowknife Bay (Vaniman
279 *et al.* 2014). Iron oxides are also found at Mawrth Vallis (Wray *et al.* 2008) and haematite was
280 identified in the floor of the Coogoon Vallis channel system (Turner *et al. submitted*) that could have
281 been transported by fluvial systems into the Oxia basin (Fawdon *et al.* 2022). As such, haematite was
282 added to the simulant.

283 Amorphous material was also detected in the JK and CB drill fines and, while the components of this
284 material are unclear, nanophase Fe^{3+} oxides such as ferrihydrite and Fe-glasses have been suggested
285 as candidate components (Dehouck *et al.* 2014). Fe-glasses have also been detected at the unnamed
286 crater in Arabia Terra, possibly forming in ancient impacts (Lai *et al.* 2019), and are therefore added
287 to the simulant. Ferrihydrite could not be sourced within the budget constraints of the project, so it
288 was substituted by amorphous iron oxyhydroxide, which has similar crystal structure.

289 Siderite, an Fe-carbonate, has also been suggested to be present alongside the phyllosilicates at Oxia
 290 Planum (Mandon *et al.* 2021; Brossier *et al.* 2022). Siderite forms in sedimentary strata laid down
 291 under anoxic conditions where Fe²⁺ is present (Lin *et al.* 2020) and is consistent with the lacustrine
 292 environment at Oxia Planum. Carbonates have also been detected in association with weathered
 293 phyllosilicates at Mawrth Vallis where weathering profiles transition from Fe/Mg phyllosilicates to Al-
 294 phyllosilicates (Bultel *et al.* 2019). Fe, Ca-rich Carbonates are suggested to account for the ~1.1-1.7
 295 µm feature in the CRISM spectra at Oxia Planum (Mandon *et al.* 2021), and a Fe/Mg-carbonates
 296 is proposed to explain the 2.52 µm feature (Mandon *et al.* 2021; Brossier *et al.* 2022). As such,
 297 siderite was added to the simulant. The resulting simulant mineralogy is summarised in Table 1.

298 Sulphates (gypsum, bassanite and anhydrite) have been proposed to occur in trace amounts in the JK
 299 and CB drill fines (4% and 1.6 wt.%, respectively); however, these sulphates are associated with
 300 Hesperian-aged surface veins that did not form in association with the phyllosilicate minerals, but
 301 were infilled at a different stage (Bibring *et al.* 2006; Ehlmann *et al.* 2011; Vaniman *et al.*
 302 2014), sulphate. Similar sulphate veining may occur at Oxia Planum; however, this simulant will only
 303 represent the regions where basaltic rocks are altered to phyllosilicates so sulphates were not added
 304 to SOPHIA.

305 **Table 1:** Minerals contained in the SOPHIA simulant

<i>Ideal Mineral</i>	<i>Amount to be added (wt. %)</i>	<i>Simulant mineralogy (Substitutes in bold)</i>	<i>Source</i>
<i>Vermiculite</i>	36	<i>Vermiculite</i>	<i>Dupree Minerals, Uganda</i>
<i>Serpentine</i>	3	<i>Serpentine</i>	<i>Taylor minerals, UK</i>
<i>Chlorite</i>	2	<i>Biotite</i>	<i>Geosupplies, Norway</i>
<i>Total phyllosilicate</i>	40		
<i>Plagioclase</i>	22	<i>Plagioclase</i>	<i>Norwegian Edelsplitt AS, Norway</i>
<i>Orthoclase</i>	3	<i>Orthoclase</i>	<i>Geosupplies, Norway</i>
<i>Pyroxene (Augite)</i>	15	<i>Pyroxene (Augite)</i>	<i>Ward Science, Norway</i>
<i>Olivine</i>	5	<i>Olivine</i>	<i>Scangrit, UK</i>
<i>Total anhydrous minerals</i>	45		
<i>Ferrihydrite</i>	1	<i>Amorphous FeOOH</i>	<i>Rowaphos</i>
<i>Fe-glass</i>	10	<i>Fe-Silicate</i>	<i>Scangrit, UK</i>
<i>Total amorphous component</i>	11		
<i>Haematite</i>	1	<i>Haematite</i>	<i>Geosupplies, UK</i>
<i>Siderite</i>	2	<i>Siderite</i>	<i>Taylor minerals, Wales</i>

306

307 2.3 Determining mineral proportions

308 As the mineral data from Oxia Planum at this point is purely orbital, there is limited ability to constrain
309 the abundance of minerals from orbit, this data is therefore incorporated with orbital data from
310 relevant sites and with *in situ* quantitative data at Gale Crater in phyllosilicate terrains. Orbital data
311 from OMEGA and TES at the unnamed crater in Arabia Terra have been used to determine the relative
312 abundances of pyroxene, Fe-glass and olivine. Using TES quantitatively can predict mineral
313 abundances to within 5 % accuracy. Here, it is used as a guide to constrain the relative amounts of
314 minerals, such that pyroxene must occur in a higher proportion to Fe-glass, and Fe-glass in a higher
315 proportion to olivine (Lai *et al.* 2019). In the Fe- phyllosilicate terrains at Mawrth Vallis, the mineral
316 abundances on average are; plagioclase 14.57 %, pyroxene 15.76 %, olivine 1.17 %, iron oxides 1.32%,
317 Fe/Mg phyllosilicates 69.11 % (Riu *et al.* 2022). These values are also used to guide the simulant
318 proportions.

319 To represent a phyllosilicate rich terrain at Oxia Planum, the simulant will contain 35 wt. %
320 phyllosilicate, since non-linear unmixing models deriving relative mineral abundances from orbital
321 data at Oxia Planum suggest an upper limit of 35 % Fe-phyllosilicates in the region (Riu *et al.* 2022). .
322 Finally, the remaining mineral proportions were constrained using quantitative XRD data from CheMin
323 on the *Curiosity* rover for similar minerals at Yellowknife Bay (Vaniman *et al.* 2014). While the
324 geological context of Oxia Planum and Gale crater are likely to be different, CheMin provides the only
325 *in situ* XRD data for Mars that can provide abundances for minerals from samples relevant to rover
326 operations. Siderite was the only mineral that was not comparable at Yellowknife Bay, and so its
327 abundance in the simulant was determined using data from Glen Torridon where it is present at 2.2
328 wt. % (Rull *et al.* 2017). The final component abundances are shown in Table 1.

329 2.4 Determining grain size for the simulant

330 As this simulant was designed to be a mineralogical simulant, its physical characteristics were not the
331 focus, however since the simulant was to be manufactured, its grain size was considered.

332 No specific grain size estimations for the phyllosilicates at Oxia Planum have been made but data from
333 the Thermal Emission Imaging System (THEMIS) instrument on Mars Odyssey suggested grain sizes of
334 either coarse sands, or a mixture of coherent rock and finer particulate sand might be anticipated in
335 the phyllosilicate terrains identified by Quantin-Nataf *et al.* (2021) (Gary-Bicas and Rogers 2021). As
336 such, grain sizes for SOPHIA were chosen based on grain sizes at Yellowknife Bay, given the possible
337 similarities between the depositional environments of the two locations. At Yellowknife Bay, the Mars
338 Hand-held Imager (MAHLI) identified grain sizes between 44 and 60 μm in diameter for the
339 phyllosilicate-bearing rocks in the Sheepbed member (Rivera-Hernández *et al.* 2019). Grain size
340 estimations are also provided by point-to-point chemical variabilities in ChemCam Laser Induced

341 Breakdown Spectroscopy (LIBS) data at Gale crater, which has suggested grains within Sheepbed
342 member mudstones are $<62.5 \mu\text{m}$, constituting a silt-clay sediment (Rivera-Hernández *et al.* 2019).
343 Analysis of MAHLI images on a pixel-by-pixel basis suggest that material at Gale crater may be
344 dominated by grains as small as $20 \mu\text{m}$ (Schieber 2018). In addition, possible similarities in grain size
345 are suggested by the scale of fracturing (Parkes Bowen *et al.* 2020). On Earth, the grain size of
346 phyllosilicate minerals typically fall into the 'clay' grain size classification, typically measuring $<2 \mu\text{m}$ in
347 diameter (Brindley 1983).

348 3 Simulant development

349 3.1 Test samples

350 Once the simulant components were identified, test samples were acquired from the suppliers (Table
351 1) to confirm their mineralogy and assess their purity prior to purchasing the required quantities. The
352 identification of any accessory mineral phases allowed the proportions of each component to be
353 adjusted to achieve the desired target simulant mineralogy. Each sample was analysed using scanning
354 electron microscopy-energy dispersive spectroscopy (SEM-EDS), which confirmed the mineralogy,
355 elemental composition and the homogeneity of the samples (see section 3.1.1). Once the mineralogy
356 of the samples was confirmed, the simulant components were ordered in bulk to make 30 kg of
357 simulant.

358 3.2 Simulant production

359 The component minerals were split, crushed, milled, and sieved at the crushing facility at The Open
360 University, UK. A rock splitter was used to split samples larger than 4 cm^3 (orthoclase, serpentine, and
361 augite). Samples were then crushed in a jaw crusher to a size of $\sim 0.5\text{-}3 \text{ mm}$ in diameter (orthoclase,
362 serpentine, augite, siderite, haematite). The platy crystal habit of biotite meant it could not be crushed
363 in the jaw crusher and was instead broken into small pieces using a rock hammer. A Retsch PM400
364 planetary ball mill was used to further crush all simulant components to a fine sand, which were then
365 sieved to $<212 \mu\text{m}$. This grain size was chosen as the time required to mill and sieve tens of kg to <60
366 μm was not feasible. The mineral components were mixed and homogenised in large mixing bowls
367 and subsampled using the coning and quartering technique (Campos and Int 2017). This
368 contamination during simulant production was contained by cleaning splitting and crushing milling
369 machinery first with decon 90 followed by acetone. Ball mill containers could be removed from the
370 crushing facility so were further solvent cleaned using dichloromethane and methanol with sonication
371 to remove a range of organics. The simulant was stored in glass jars and the lids sealed with foil to
372 prevent contamination.

373 4 Simulant Characterisation

374 The following Characterisation data is available at (10.21954/ou.rd.22219903). Photographs of the
375 mineral components in their supplied, milled and sieved states and at $\times 50$ magnification when
376 polished and embedded in epoxy are provided in Table S1.

377 4.1 Scanning electron microscopy - energy dispersive spectroscopy (SEM-EDS)

378 SEM-EDS analysis was conducted on test samples ($\sim 3 \times 3$ cm) to confirm their mineralogy, elemental
379 composition and homogeneity. The samples were embedded into epoxy resin blocks of 4 cm diameter,
380 polished flat and carbon coated before analysis. These samples were analysed using a FEI Quanta 200
381 3D scanning Electron Microscope (SEM), equipped with an Oxford Instruments Energy Dispersive X-
382 ray detector (EDS), using a 20 kV electron beam. The working distance was 15 mm. EDS chemical
383 mapping was coupled with backscatter electron (BSE) imaging. Bulk chemical data for the test samples
384 were collected using randomised large-area mapping, whereby multiple maps were acquired over the
385 surface at regularly intervals and collated. This provided a detailed composition representative of the
386 test sample as a whole, negating the sampling bias associated with choosing a specific location to map.
387 This provided an overview of the sample mineralogy and supplied semi-quantitative information on
388 the proportion of mineral phases in the samples. SEM-EDS results for components are shown in Table
389 2. To assess the simulant's bulk chemistry, an SEM-EDS map was taken using the parameters described
390 above. The bulk chemical composition based on large area chemical mapping is shown in Table 3. The
391 map with identified minerals is shown in Figure 2.

Table 2: SEM-EDS data for the component minerals for the simulant (in the test samples), identifying dominant mineral phases and contaminant mineral phases.

	Simulant component	Mineral phase	%	Oxide percentages in each mineral phase											Cr ₂ O ₃	CO ₂
				MgO	Al ₂ O ₃	SiO ₂	K ₂ O	FeO	Fe ₂ O ₃	Na ₂ O	CaO	Ti ₂ O ₂	MnO			
1	Vermiculite	Fe/mg-vermiculite	91	20.53	21.75	43.45	5.53	7.21	-	-	0.45	1.11	-	-	-	
		Al-vermiculite	6	12.33	59.56	21.03	2.18	4.62	-	-	-	-	-	-	-	
		Mg-vermiculite	3	29.30	12.75	44.23	-	7.13	-	-	-	-	-	-	-	
2	Anorthosite	labradorite	72	-	26.35	56.87	0.58	-	0.32	6.01	9.87	-	-	-	-	
		Oligoclase	25	-	25.30	62.28	1.46	0.19	-	8.32	2.46	-	-	-	-	
		Sanidine	2	-	34.66	51.91	9.15	0.87	-	1.89	1.52	-	-	-	-	
		Albite	<1	-	19.90	69.17	-	-	-	10.93	-	-	-	-	-	
		Bytownite*	<1	-	32.68	47.69	5.63	0.97	-	0.89	12.15	-	-	-	-	
		Unknown	<1	11.92	24.59	42.76	0.73	16.03	-	-	3.96	-	-	-	-	
		Sanidine	<1	3.88	32.62	48.12	9.73	4.54	-	-	1.12	-	-	-	-	
3	Orthoclase	Orthoclase	92	-	18.80	64.93	14.53	-	-	1.74	-	-	-	-	-	
		Plagioclase	8	-	21.34	67.26	1.56	-	-	9.85	-	-	-	-	-	
4	Haematite	Haematite	87	-	-	0.01	-	-	-	98.99	-	-	-	-	-	
		Haematite	7	-	-	2.79	-	-	-	97.22	-	-	-	-	-	
		Macaulayite*	6	-	14.28	9.21	-	76.51	-	-	-	-	-	-	-	
5	Siderite	Siderite	99	-	-	-	-	52.14	-	-	0.74	-	1.32	-	45.80	
		Calcite	1	-	-	-	-	3.37	-	-	44.9	-	0.37	-	51.33	
6	Amorphous β-FeOOH	Amorphous β-FeOOH	100	-	-	-	-	-	100	-	-	-	-	-	-	
7	Serpentine	Lizardite	75	40.59	3.13	49.31	-	6.79	-	-	-	-	-	-	-	
		Serpentine	34	44.60	-	47.3	-	-	-	-	0.35	0.22	-	-	-	
		Olivine	1	28.34	-	28.84	-	42.83	-	-	-	-	-	-	-	
		Tremolite*	<1	31.25	-	54.38	-	4.41	-	-	9.96	-	-	-	-	
8	Pyroxene	Pyroxene (omphacite)	47	-	32.18	51.77	5.48	-	-	7.83	2.74	-	-	-	-	
		Pyroxene (augite)	39	11.48	-	51.90	-	9.40	-	-	23.99	3.28	-	-	-	
		Pyroxene (pigeonite)	9	29.92	-	34.41	0.01	34.17	-	-	0.06	0.01	0.07	-	-	

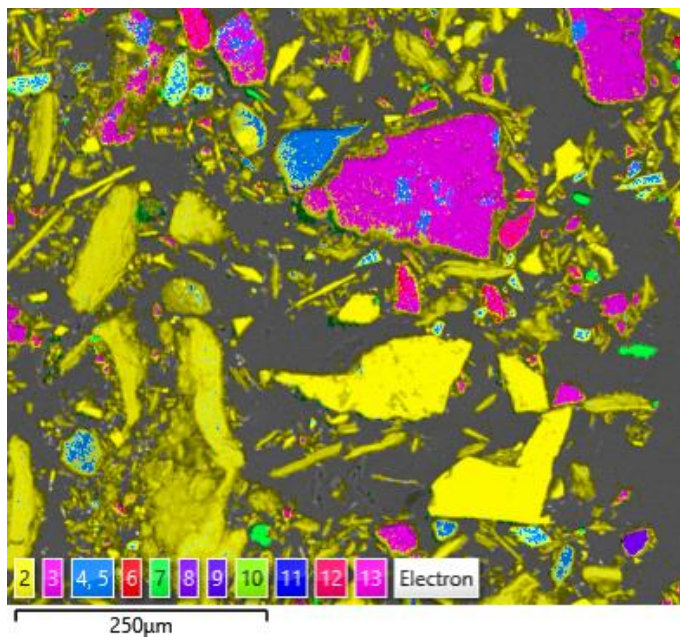
		<i>Amphibole (kaersuite)*</i>	5	6.41	17.12	47.47	1.52	7.80	-	5.08	12.38	-	-	-	-
		<i>Orthoclase</i>	<1	-	19.94	62.84	12.74	0.90	-	2.53	0.75	0.31	-	-	-
9	<i>Fe-silicate</i>	<i>Fe-silicate</i>	94	-	4.52	38.84	0.58	51.43	-	-	4.25	0.38	-	-	-
		<i>Fe,Ca-Silicate</i>	6	3.23	4.28	33.01		43.34	-	-	16.13				-
10	<i>Olivine</i>	<i>Mg rich olivine (chrysolite)</i>	98	50.17	-	42.27	-	7.57	-	-	-	-	-	-	-
		<i>Hornblende*</i>	1	22.87	4.77	53.79	-	3.84	-	-	12.70	-	-	2.03	-
		<i>Pyroxene (enstatite)</i>	<1	33.18	-	63.36	-	3.47	-	-	-	-	-	-	-
11	<i>Biotite</i>	<i>Biotite</i>	99	7.52	16.14	39.17	9.61	24.21	-	-	0.08	2.78	0.49	-	-
		<i>Biotite</i>	<1	6.98	18.57	41.59	5.08	25.33	-	-	0.64	1.81	-	-	-
		<i>Unknown</i>	<1	2.14	27.43	49.52	10.5	9.1	-	-	1.32	-	-	-	-

394 *The identity of these minerals is speculative as they could not be conclusively determined. They are present in small enough amounts as to not affect the overall mineralogy
395 of the simulant.

396 **Table 3:** Bulk wt.% oxides based on the SEM-EDS and XRF analysis of the bulk simulant.

Oxide	Calculated chemistry based on mineral input	XRF %	SEM-EDS %
SiO ₂	48.28	46.42	48.16
TiO ₂	0.64	0.51	0.69
Al ₂ O ₃	17.96	11.42	14.13
Fe ₂ O ₃ /FeO	12.89	13.49	12.82
MnO	0.03	0.14	n/a
MgO	8.28	13.77	15.79
CaO	4.08	6.07	4.95
Na ₂ O	2.05	1.73	n/a
K ₂ O	2.84	2.98	3.46
P ₂ O ₅	n/a	0.051	n/a
SO ₃	n/a	0.094	n/a
Lol		3.73	n/a

397



398

399 **Figure 2:** False-colour SEM-EDS map showing the different components in the bulk simulant based on the
 400 elements detected. (2) Vermiculite (3) Oligoclase (4, 5) Pyroxene (6) Orthoclase (7) Haematite/Siderite/
 401 Amorphous β-FeOOH (9) Olivine.
 402

403 4.2 Raman Spectroscopy

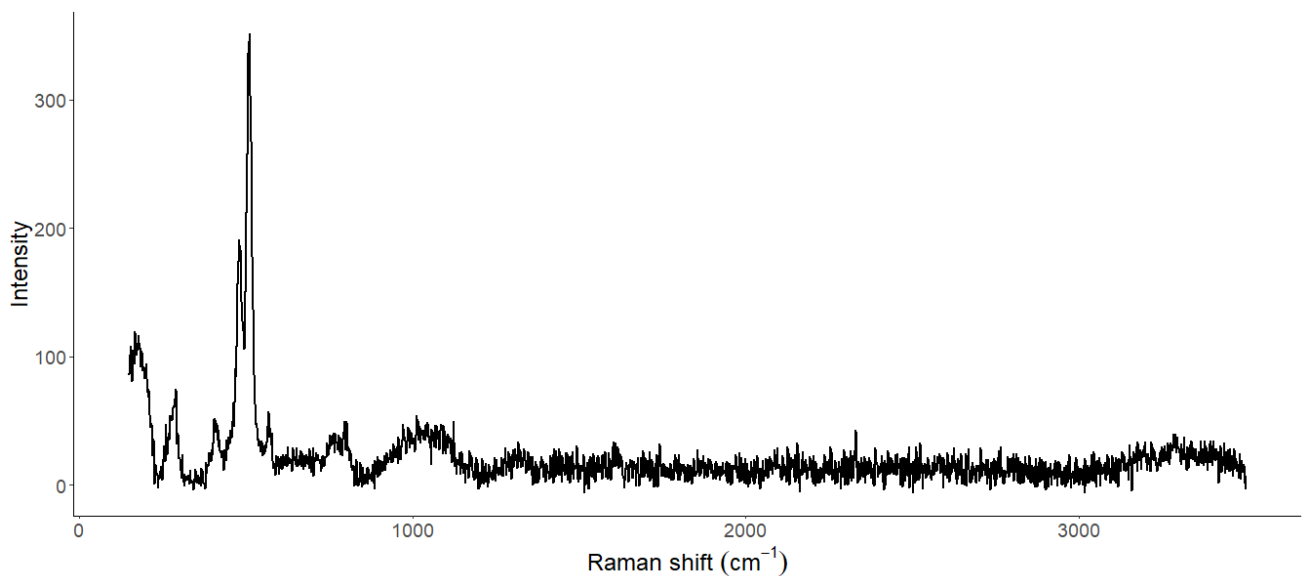
404 Raman spectroscopy was used to confirm the mineral phases present in each of the test samples. A
 405 Horiba Jobin-Yvon HR800 spectrometer was used, which was equipped with an Ar ion laser (of
 406 wavelength 514 nm) as the excitation source. This wavelength was used to ensure the results were
 407 comparable to the Raman Laser Spectrometer (RLS) instrument aboard the *Rosalind Franklin* rover
 408 (Rull *et al.* 2017). To prevent the laser burning the samples and to prevent thermally-induced
 409 fluorescence, a 10% neutral density filter (N.D.) was used, reducing the power output at the microscope

410 objective. A $\times 10$ microscope objective was used to achieve a spot size of $2.5\ \mu\text{m}$, while a 600
411 grooves/mm diffraction grating gave a spectral resolution of $1.5\ \text{cm}^{-1}$. A montage ($1000 \times 1000\ \mu\text{m}$)
412 was acquired, and a map consisting of 100 spectra was collected for this area. Raman spectra were
413 analysed using the *Labspec 6* software by applying a baseline correction and then a peak fitting
414 function using a Gaussian-Lorentzian profile to identify peak positions and peak widths (taken at full
415 width half maxima). Minerals were identified using characteristic bands reported in the literature and
416 the RRUFF database (Lafuente *et al.* 2015). The characteristic peaks for the phases identified are
417 presented in Table 4. The Raman spectrum for the bulk simulant is shown in Figure 3.

418 **Table 4:** Confirmation of mineral composition by Raman spectroscopy

Sample	Mineral phase	Experimental peaks (Raman shift (cm ⁻¹))	Reference
Vermiculite	Vermiculite	195, 360, 548, 682	Lafuente et al., (2015)
Serpentine	Lizardite	223, 298, 408, 678	Lafuente et al., (2015)
	Chrysotile	277, 414, 690, 1033	Lafuente et al. (2015)
Biotite	Biotite	231, 389, 686	Wang et al., (2015)
	Phlogopite	234, 303, 389, 697, 827, 863	Wang et al., (2015)
Synthetic Fe-silicate		920	
Plagioclase (Anorthosite)	Labradorite	207, 298, 424, 488, 510	Lafuente et al., (2015)
	Oligoclase	209, 282, 488, 515, 770, 825, 1126	Lafuente et al., (2015)
Orthoclase	Orthoclase	166, 267, 288, 460, 477, 518, 756, 816, 1142	Lafuente et al., (2015)
Pyroxene		327, 403, 488, 507, 658, 819, 857, 1013	
Olivine	Olivine	157, 347, 400, 637, 816, 855, 915, 945, 1114, 1602	Berlanga et al., (2019)
	Pyroxene	234, 347, 678, 1017, 1602	
Haematite	Haematite	225, 285, 414, 469, 617, 1322	Lafuente et al., (2015)
Amorphous β-FeOOH	Amorphous β-FeOOH	160, 218, 284, 893	n/a
Siderite	Siderite	187, 283, 1087	Lafuente et al. 2015)

419



420

421 **Figure 3** Raman spectrum for the SOPHIA simulant showing peaks at 186 and 512 cm⁻¹ associated with
 422 phyllosilicates; florescence from 1000 cm⁻¹ to 3500 cm⁻¹ is also due to the presence of phyllosilicates, removed
 423 with baseline correction.

424

425 4.3 Electron microprobe analysis (EMPA)

426 EMPA was used to obtain quantitative chemical data for the dominant mineral phases in each test
427 sample. A Cameca SX100 microprobe equipped with five spectrometers was used for analysis. Samples
428 were analysed using a 10 µm beam diameter, using a 20 kV accelerating voltage and 20 nA current.
429 The probe was programmed to measure elemental concentrations of K, Mn, Ca, Fe, Si, Mg, Cr, Ti, Al,
430 Si as these elements were shown to be present in the samples with SEM-EDS. EMPA can determine
431 the chemical composition with higher sensitivity than SEM-EDS with a spectral resolution of roughly
432 an order of magnitude higher. Standards within the software were used to calibrate the probe prior
433 to the unknown compositions being analysed. The vermiculite samples could not be analysed by EMPA
434 because of the challenges associated with preparing polished blocks(Fitzpatrick 1984), and because of
435 damage to the sample caused by the interaction with the electron beam; these are recognised
436 challenges associated with analysing phyllosilicate minerals (Biroň *et al.* 1999). The EMPA results are
437 shown in Table 5.

438 **Table 5:** Electron microprobe characterisation of the simulant components

Component	Mineral Phase	K₂O	MnO	FeO	Fe₂O₃	CaO	Na₂O	Al₂O₃	SiO₂	MgO	Cr₂O₃	TiO₂	P₂O₅	SO₂	H₂O
<i>Vermiculite</i>	<i>Vermiculite*</i>	2.5	0.02	3.13	-	0.02	0.43	5.77	23.67	13.80	0.01	0.55	-	0.28	49.63
<i>Plagioclase</i>	<i>Plagioclase</i>	4.29	-	0.39	-	3.27	6.13	27.45	58.06	0.20	-	0.01	0.02	0.01	-
<i>Orthoclase</i>	<i>Orthoclase</i>	16.3	-	0.03	-	0.04	0.65	18.45	64.48	-	-	0.03	-	-	-
<i>Haematite</i>	<i>Haematite</i>	-	-	-	98.67	-	-	-	-	-	-	-	-	-	-
<i>Siderite</i>	<i>Siderite</i>	-	2.03	92.41	-	1.44	-	4.00	0.05	0	-	-	-	-	-
<i>Amorphous β-FeOOH</i>	<i>Iron Oxide</i>	-	-	-	72.00	-	-	-	-	-	-	-	-	-	28.00
<i>Serpentine</i>	<i>Lizardite</i>	0.01	0.24	7.22	-	0.09	-	0.86	40.1`	36.05	0.02	0.04	0.01	0.03	15.32
	<i>Tremolite</i>	0.02	0.09	3.21	-	11.92	0.63	3.48	53.34	22.75	0.46	0.35	0.02	0.01	3.72
<i>Pyroxene</i>	<i>Omphacite</i>	4.02	0.01	0.47	-	2.85	8.08	26.3	58.09	0.03	-	0.09	-	-	-
	<i>Augite</i>	-	0.17	8.72	-	22.85	0.65	7.67	46.2	10.78	0.05	2.86	0.03	-	-
	<i>Orthoclase</i>	10.8	-	0.14	-	0.72	4.12	19.37	64.75	-	-	0.08	-	-	-
	<i>Iron silicate</i>	9.12	0.19	19.85	-	0.90	0.6	13.62	37.6	11.3	0.01	6.79	-	0.01	-
	<i>Plagioclase</i>	0.56	0.01	0.30	-	0.49	13.08	27.79	57.31	0.05	-	0.03	0.37	-	-
	<i>Fe-Ti Oxide</i>	0.01	1.23	73.1	-	0.06	-	1.79	0.16	0.44	0.24	22.98	-	-	-
<i>Fe-silicate</i>	<i>Fe-silicate</i>	0.58	0.55	48.39	-	4.52	0.61	4.75	38.05	1.26	0.25	0.26	0.22	0.57	0.58
<i>Olivine</i>	<i>Olivine</i>	-	-	7.19	-	-	-	-	41.15	51.25	-	-	-	-	-
	<i>Olivine</i>	-	-	6.17	-	-	-	-	46.76	46.85	-	-	-	-	-
	<i>Serpentine</i>	0.01	0.01	2.77	-	0.01	-	16.33	38.29	38.64	3.99	0.03	0.01	0.02	-
<i>Biotite</i>	<i>Biotite</i>	9.55	0.48	24.18	0.03	0.08	15.22	36.44	7.48	-	2.53	-	0.05	3.95	9.55

439 *Vermiculite could not be analysed accurately because the samples could not be polished without plucking out the soft grains, resulting in an irregular surface
440 that meant the probe was difficult to focus.

441 **4.4 X-Ray Florescence Spectroscopy**

442 X-Ray Florescence Spectroscopy (XRF) was used to provide quantitative chemical data for the bulk
443 simulant. This was conducted at the University of Leicester, UK, using a Rigaku ZSX PrimusIV WD-XRF.
444 Homogenised samples were crushed to <53 µm in diameter, and ignited powders were used to
445 prepare fused silica beads on which major elements were determined, with a sample to flux ratio of
446 1:7.5 66% Li metaborate: 34% Li tetraborate flux. Results are quoted as component oxide weight
447 percent (wt.%) and recalculated to include Loss on Ignition (LOI). A PANalytical Axios Advanced WD-
448 XRF was used to analyse trace elements. Samples were made into 32 mm diameter powder pellets
449 produced by mixing 10 g of ground simulant with 20-25 drops of a 7% PVA solution (Moviol 8-88) and
450 pressed at 152 MPa. The XRF data is shown in Table 3. In addition, the XRF results can be used to
451 assess the Chemical Index of Alteration (CIA) for the simulant to understand the type of alteration
452 environment (Nesbitt and Young 1982) it represents on Mars (Equation 1), giving a CIA of 51 for the
453 simulant, usually associated with an environment bordering between an open and closed system.

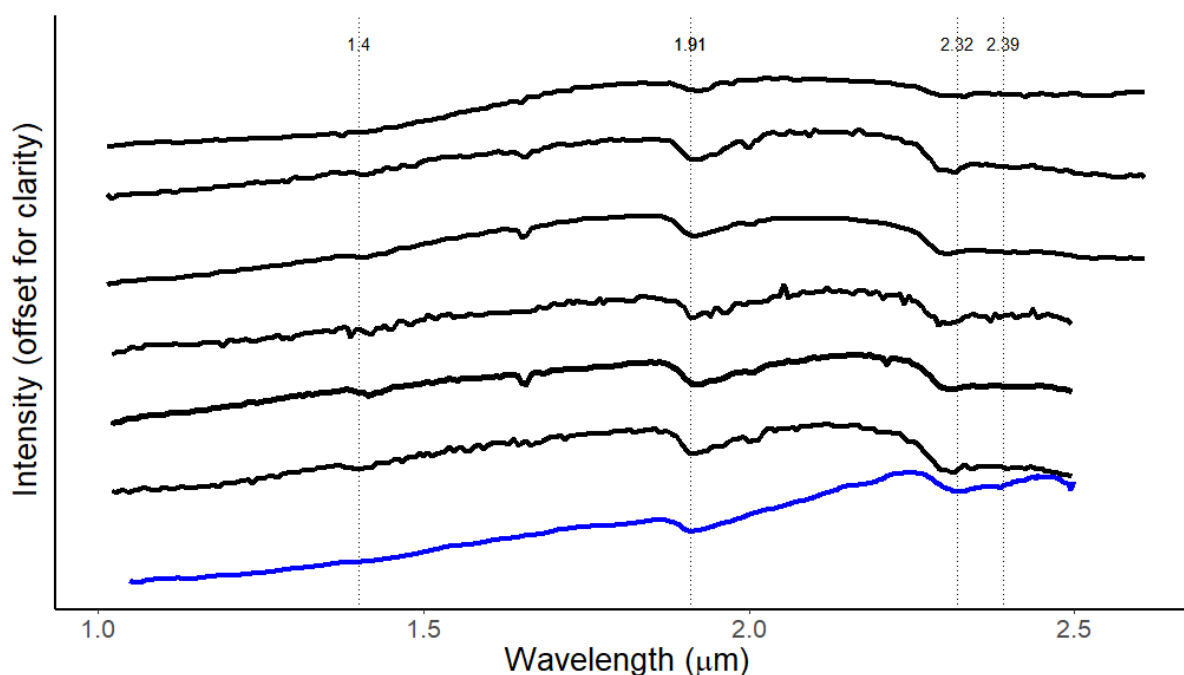
454
$$4 \quad CIA = 100 \times \frac{Al_2O_3}{Al_2O_3 + CaO + Na_2O + K_2O} \text{ (Equation 1)}$$

455

456 **4.5 Near-IR Spectroscopy**

457 Near-IR spectroscopy was used to characterise the bulk simulant, which allowed the data to be
458 compared to the CRISM spectra of Oxia Planum and can be used to compare the spectral similarities
459 between the simulant and Oxia Planum from orbit before the *Rosalind Franklin* rover lands. The data
460 can also be used to compare to the rover's ISEM (Korablev *et al.* 2017), MicrOmega (Bibring *et al.*
461 2017) and Ma_MISS (contact in the drill hole) instruments (De Sanctis *et al.* 2022). Near-IR spectra
462 were obtained using an Avantes AvaSpec-NIR512-2.5-HSC-EVO (with thermoelectric cooling) using a
463 100 lines/mm grating with a 1350-2500 nm usable spectral range. These parameters provide a spatial
464 resolution of 7.2-50 µm. The spectrometer was used in conjunction with AvaSoft 8 software where
465 calibration of the instrument and smoothing of data was performed. Calibration was performed using
466 white and dark references before acquiring spectra. Once spectra were acquired, standard correction
467 smoothing was performed as a standard procedure. An example spectrum for the simulant is shown
468 in Figure 4 (blue spectrum). The CRISM spectra at Oxia Planum are also shown for comparison (black
469 spectra).

470



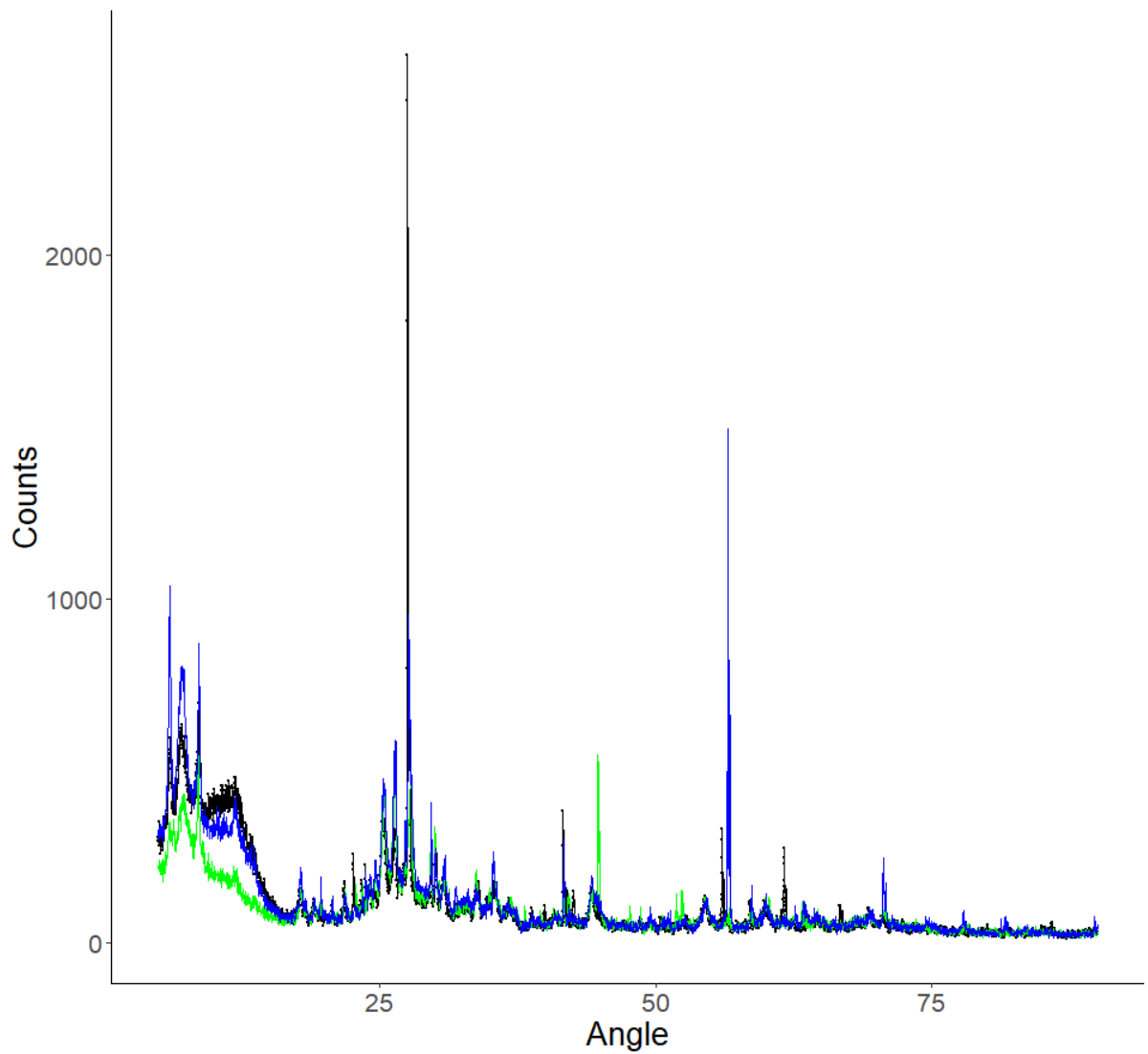
471

472 **Figure 4:** In black, the Near-IR spectrum for the bulk simulant shows absorption bands shown using dotted lines
 473 at 1.92 and 2.3 μm associated with hydration, and absorption at 2.5 μm is possibly due to the mixing of
 474 vermiculite with siderite and/or serpentine (blue spectrum). Three Near-IR spectra from Oxia Planum taken by
 475 the CRISM instrument show absorption at 1.9 μm and 2.3 μm (shown in black; Turner and Bridges 2017; Mandon
 476 et al. 2021).

477 4.6 X-Ray Diffraction (XRD)

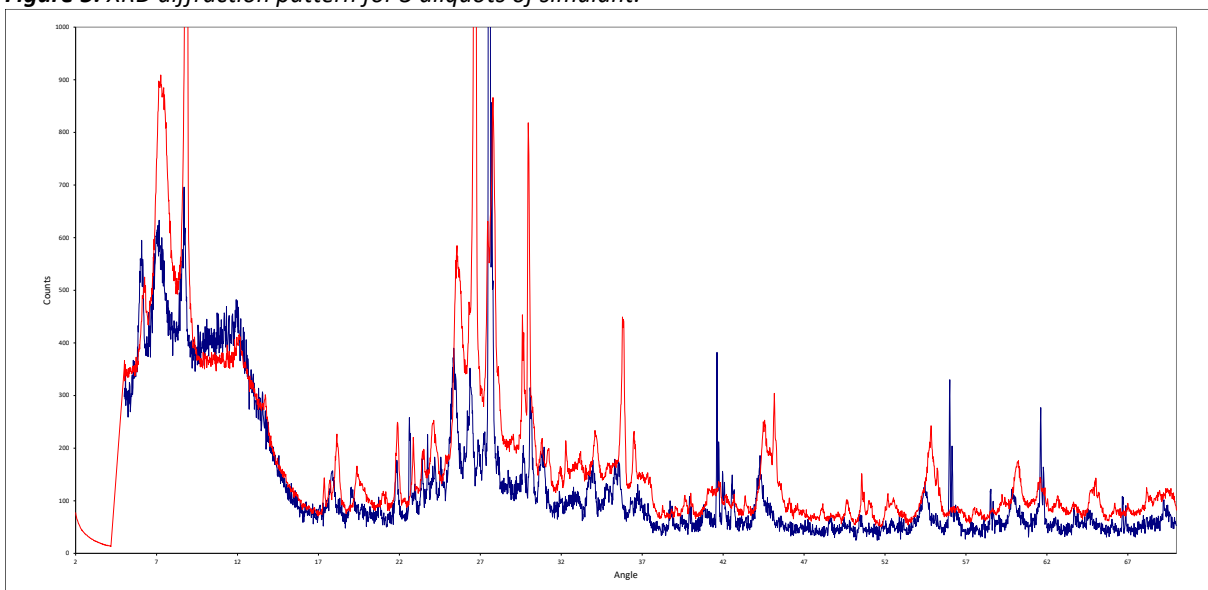
478 X-Ray diffraction (XRD) was used to assess the mineralogy of the bulk simulant, quantify its
 479 composition and verify its homogeneity. Qualitative analysis was conducted at the Open University,
 480 using a Siemens D5000 taking measurements from 5 to 90 degrees in 0.02 step. This was repeated on
 481 three aliquots of simulant to ensure the simulant was homogenous (Figure 5).

482 XRD analysis was conducted on each of the component minerals and the diffraction patterns from this
 483 analysis were combined using FULLPAT software (Chipera and Bish 2002) to provide a theoretical
 484 diffraction pattern for the combined simulant. This analysis is compared to the XRD analysis for the
 485 bulk spectra in Figure 6. Rietveld refinement was also applied to the diffraction pattern to acquire
 486 quantitative data (Table S2). In addition, XRD analysis, to fully quantify the amorphous material, was
 487 run at Sheffield Hallam University using a PANalytical Empyrean X-ray Powder Diffractometer. A 3.45
 488 w% of NIST SRM 640e silicon XRD standard powder was added to the simulant for comparison to the
 489 simulant's amorphous content.



490

491 **Figure 5:** XRD diffraction pattern for 3 aliquots of simulant.



492

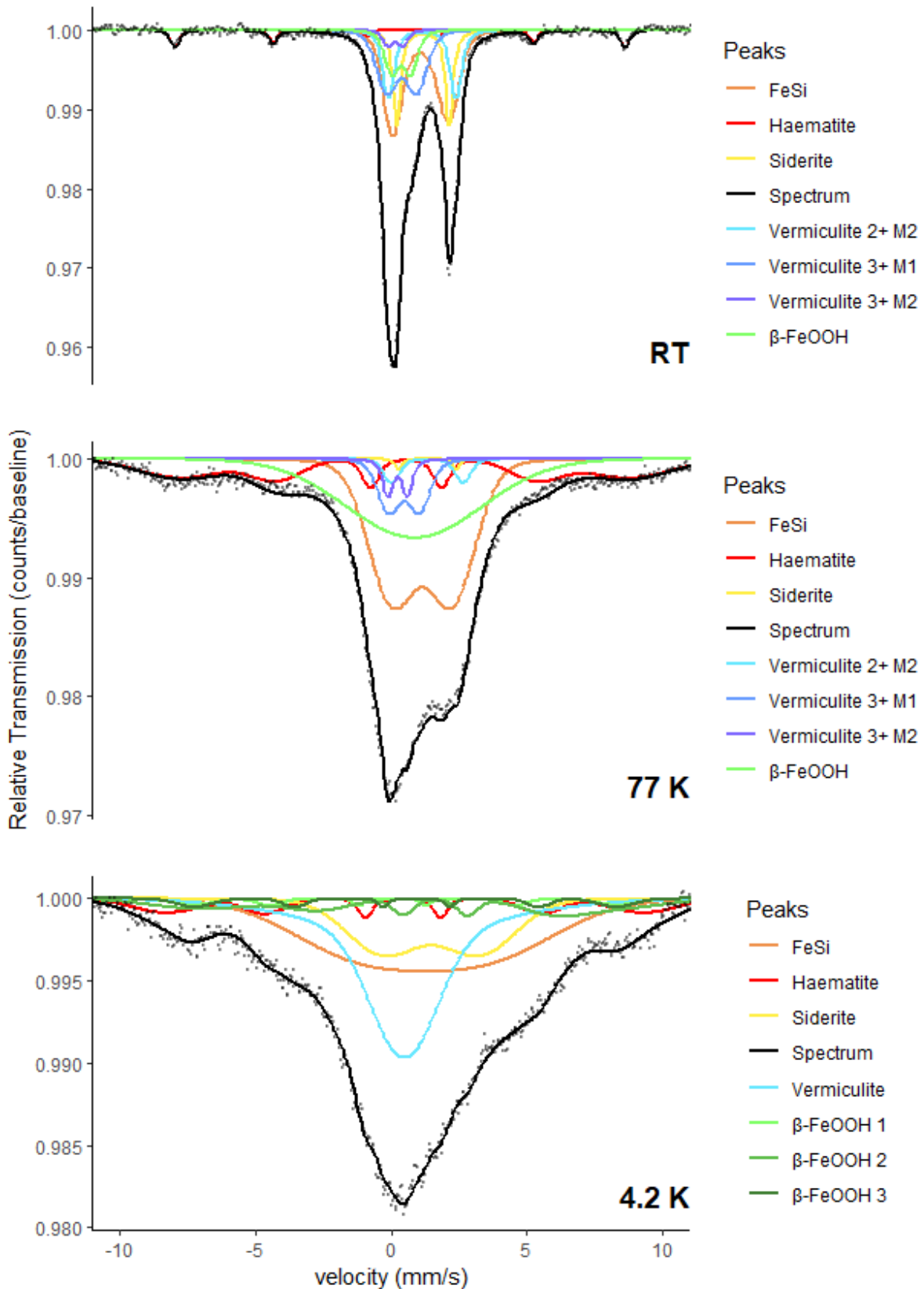
493 **Figure 6:** XRD diffraction patterns showing the analysis of the bulk simulant (blue) and the theoretical pattern
 494 for the sum of the components (red).

495

496 4.7 Mössbauer Spectroscopy

497 Mössbauer spectroscopy (e.g. Gütlich and Schröder 2012) was used to analyse the Fe²⁺/Fe³⁺ ratio of
498 the bulk simulant by characterising the mineralogy of the Fe-bearing minerals in the simulant
499 (vermiculite, siderite, amorphous β-FeOOH, haematite and Fe-silicate). Mössbauer was used onboard
500 the Mars Exploration Rovers (MER) *Spirit* and *Opportunity* (Klingelhöfer *et al.* 2003) that analysed
501 surfaces materials in Gusev crater and at Meridiani Planum (Klingelhöfer *et al.* 2004; Morris *et al.*
502 2006a, 2006b, 2019).

503 Mössbauer analysis of the simulant was conducted at the Mössbauer Spectroscopy laboratory for
504 Earth and Environment (MoSEE) at the University of Stirling. Aliquots of the bulk simulant (116 mg),
505 and the individual components Fe-silicate glass (115 mg), β-FeOOH (103 mg) and vermiculite (110 mg)
506 were placed into acrylic glass sample holders (~1.3 cm circular diameter) and measured at three
507 different temperatures (room temperature, 77 K, 4.2 K) with a standard transmission spectrometer
508 (Wissel, Germany) attached to a closed-cycle helium gas cryostat (DRY ICE 4K, ICEoxford, UK). The
509 instrument uses 14.4 keV gamma radiation emitted by a ⁵⁷Co in Rh matrix source in constant
510 acceleration mode (triangular waveform). Spectra were calibrated against a spectrum of α-iron foil
511 (25 μm thickness) at room temperature. Spectra were evaluated with Recoil (Ottawa, Canada) using
512 the Voigt-based fitting routine (Rancourt and Ping 1991). No f-factor correction was applied.
513 Mössbauer spectra are shown in Figure 7 and Mössbauer parameters are listed in Table S3.



514

515 **Figure 7:** Mössbauer spectra for simulant and iron containing components. A – Spectrum measured at room
 516 temperature. B – Spectrum measured at 77 K. C – Spectrum measured at 4.2 K.

517

518 4.8 Physical and mechanical properties

519 Characterisation of the simulant's physical properties was conducted by K4soils (K4 Soils testing
520 laboratory, Watford, UK), using 400 g of the homogenised simulant with a size fraction of $\leq 212 \mu\text{m}$.

521 The bulk density (ρ) of the simulant was calculated by performing linear measurements (Hvorslev,
522 1970) and was calculated using Equation (2)

$$523 \quad \rho = \frac{400 m}{\pi D^2 L} \text{ (Equation 2)}$$

524 where, m is mass (kg), D is mean diameter of cylinder (m), and L is mean length (m). Bulk density was
525 found to be 1150 kg m^{-3} .

526 The simulant's particle density (ρ_s) was measured using the small pycnometer method and was found
527 to be 2.91 kg m^{-3} .

528 Porosity was calculated using the bulk density and particle density measurements by determining
529 specific gravity (G_s) and void ratio (e) using Equations (3) – (5):

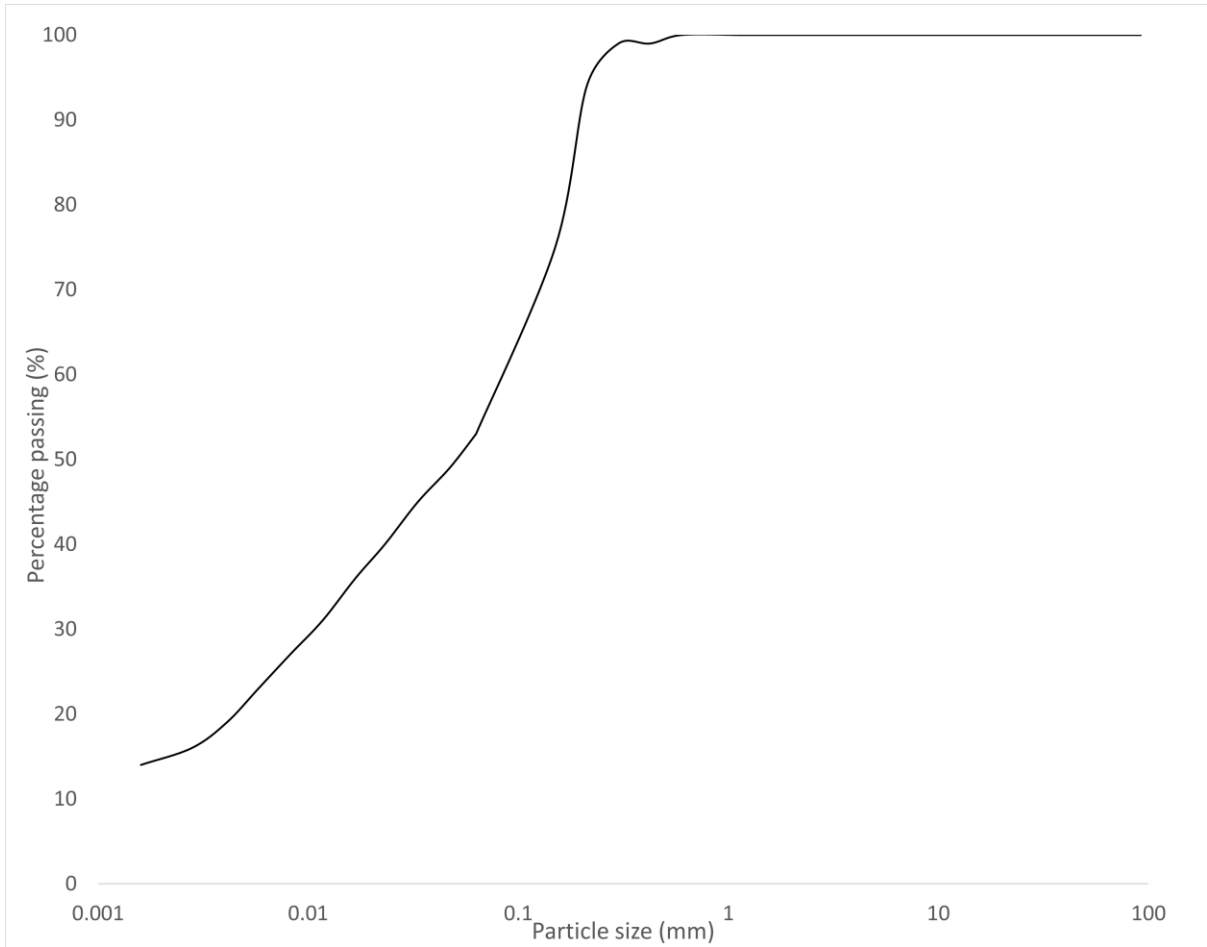
$$530 \quad G_s = \frac{\rho_s}{\rho_w} \text{ (Equation 3)}$$

$$531 \quad e = G_s \frac{\rho_w}{\rho} - 1 \text{ (Equation 4)}$$

$$532 \quad n = \frac{e}{(1+e)} \text{ (Equation 5)}$$

533 where, ρ_s is the particle density and ρ_w is the density of water (1000 kg m^{-3}), and ρ is the bulk density.
534 The simulant porosity was 61%.

535 The grain size distribution across 100 g of the bulk simulant was characterised using the GSB BS1377
536 method of small pycnometer sieving (Hvorslev *et al.*, 1949) and depositing the sedimented material.
537 The grain size fractions are 46.9 % sand (0.06-2.0 mm), 38.4 silt (3.9–62.5 μm) and 14.7 % clay (0.98–
538 3.9 μm) (Figure 8).

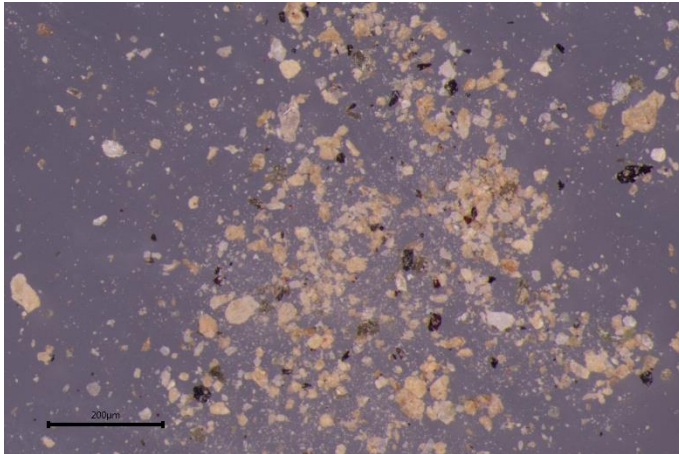


539

540 **Figure 8:** Grain size distribution of the bulk simulant

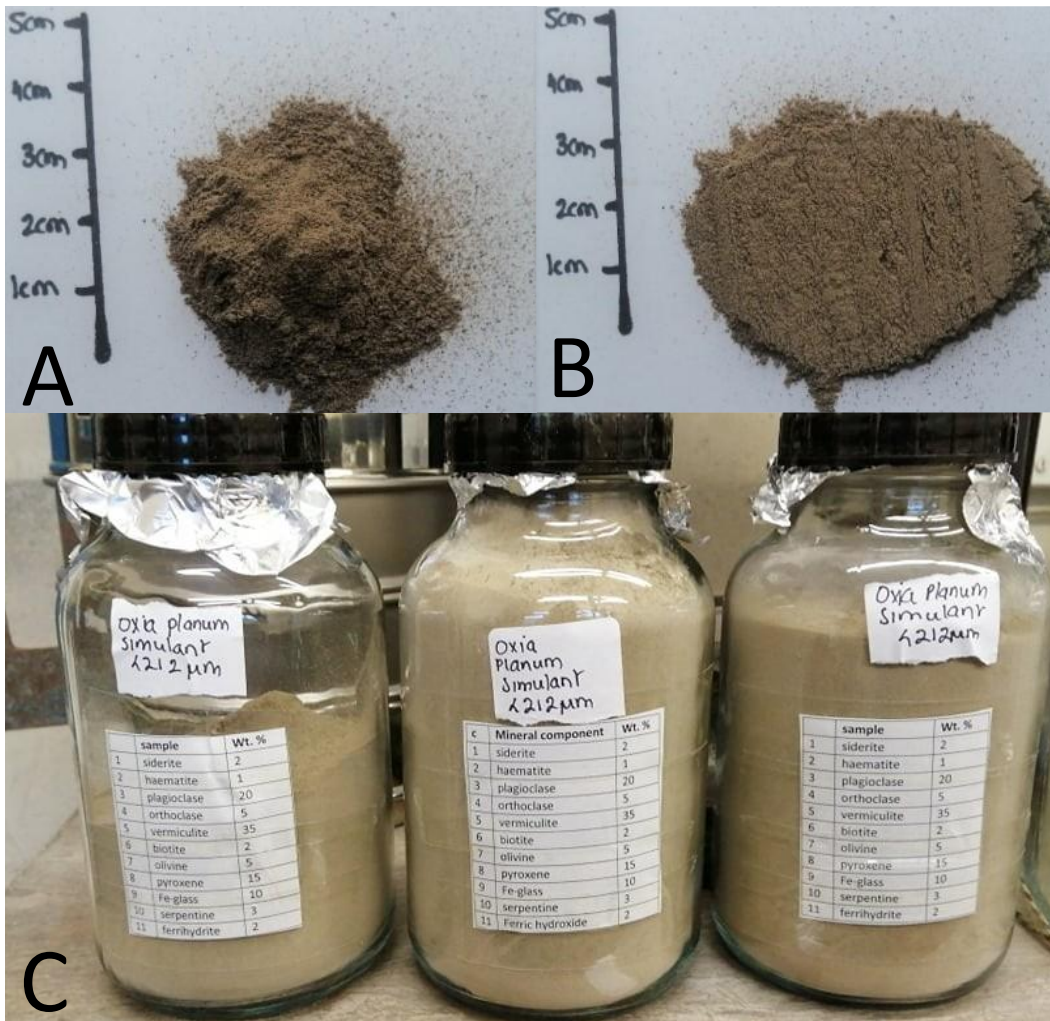
541 **4.9 Optical microscopy**

542 A Leica WILD MZ8 light microscope was used to image the bulk simulant (Figure 9). While the simulant
 543 was too fine to resolve the grain shape, expected grain shapes, based on the crystal structures of the
 544 minerals, are listed in Table S4 . Photographs of the simulant heaped and smoothed out are shown in
 545 Figure 10. These can be used in comparison to micro imager data from Mars such as Mars Hand Lens
 546 Imager (MAHLI) on *Curiosity* (Edgett *et al.* 2012) or at Oxia Planum from CLUPI (Josset *et al.* 2017),
 547 which may observe piles of drill tailings.



548

549 **Figure 9:** Optical image of the simulant showing a range of grain shapes and sizes. A dark background is used to
 550 show up light coloured minerals. Dark grains include biotite, pyroxene, Fe-silicate, red grains include siderite,
 551 haematite and FeOOH while light grains dominate the simulant including vermiculite, orthoclase plagioclase,
 552 serpentine.
 553



554

555 **Figure 10:** Photos of bulk simulant: (A) piled simulant material; (B) simulant material when levelled out,
 556 comparable with drill tailings or after being crossed with a sample flattening blade by the Rosalind Franklin rover;
 557 (C) simulant storage at the Open University in glass jars with foil covered lids to prevent contamination.

58 5 Discussion

559 The purpose of this study was to create a mineralogical simulant to represent the local mineralogy of
560 Oxia Planum at a scale that is relevant to rover operations, which can then be used to facilitate
561 analogue experimentation and support analysis of future rover data. The simulant design has been
562 achieved by combining all the available remote sensing data with the most representative mineral
563 assemblages identified from other rover exploration sites.

564 5.1 Simulant composition

565 5.1.1 Individual component analysis

566 SEM-EDS, EMPA and Raman spectroscopy were conducted to confirm the mineralogy and identify any
567 accessory minerals present in the components, to ensure the bulk mineralogy of the final simulant
568 reflected the abundances identified in Table 4.

569 SEM-EDS and EMPA analysis showed that test samples had more than one member of the same
570 mineral group present. For example, pyroxene consisted of augite, pigeonite and omphacite, and
571 plagioclase contained oligoclase and labradorite. In addition, the vermiculite sample contained Al-
572 vermiculite in addition to Mg and Fe-vermiculite. However, the simulant design did not specify a
573 member of the group, and the semi-quantitative data could be used to accommodate variations from
574 the ideal composition. For some mineral components, accessory minerals were important
575 contributors to the simulant; for example, orthoclase contained plagioclase (8 %) that was also
576 required, and serpentine contained trace amounts of olivine (<1%). Negligible amounts of accessory
577 minerals including hornblende, amphibole, Ca-silicate, calcite, macaulayite, bytownite and albite were
578 also detected in components (Table 2). Accessory minerals are expected to total to <3 % of the
579 simulant, so have no major effect on the overall mineralogy. Combining the simulant components in
580 the amounts as proposed (Table 1) therefore results in an appropriate simulant mineralogy (Table 3).

581 5.1.2 Simulant Mineralogy

582 The simulant mineralogy was analysed using XRD, Near-IR spectroscopy and Mössbauer spectroscopy.
583 Qualitative XRD from 3 aliquots of simulant shows the mineralogy of the simulant is homogenous with
584 slight variations in peak intensities being accounted for by the crystal structures of the minerals.

585 Semi-quantitative analysis of these diffraction patterns was conducted by comparison to a
586 constructed diffraction pattern based on the combined diffraction patterns of the components and
587 the amounts that they were present in, using the FULLPAT software. This theoretical diffraction
588 pattern is on good agreement with the simulant pattern suggesting the proportions in the simulant
589 are as expected. Quantitative XRD was attempted using Rietveld refinement analysis and by analysis

590 of the amorphous material. These methods produced very variable results (Table S2). This was due to
591 a range of issues: analysing mixtures of minerals can be difficult due to overlapping and broad
592 reflections, poor availability of pure standards and variation in preferred orientation to name but a
593 few (Bish and Chipera 1987). XRD analysis of phyllosilicates can be difficult because of similarities
594 between their a and b unit cell dimensions; however, the c unit cell dimension differs between
595 phyllosilicate minerals. Consequently, basal peaks (00l) are used as the primary method for
596 phyllosilicate identification because these are the most defined (Moore and Reynolds 1997). In the
597 case of a biotite and vermiculite mixture, the more defined peak for biotite leads to over or
598 underrepresentation of the relative amounts of the two. This is also true in the case of the diffuse
599 peaks for amorphous material where the inability to define a reliable identifying peak makes it difficult
600 to quantify the material. While XRD is not onboard the *Rosalind Franklin* rover, the difficulty in
601 quantifying the minerals in vermiculite-rich complex mixtures may be an issue for future missions.

602 Mössbauer analysis indicates that the $\text{Fe}^{3+}/\text{Fe}_{\text{total}}$ ratio is 0.40, making the simulant dominated by Fe^{2+} ,
603 which does not correlate to enrichment in Fe^{3+} observed for much of the phyllosilicate terrain at Oxia
604 Planum. However, the orbital data for Oxia Planum is not fully quantitative and many factors including
605 dust cover and grain size could mean ground truths vary. In addition, controlling the $\text{Fe}^{2+}/\text{Fe}^{3+}$ ratio
606 was not one of the requirements of the simulant design. Future work could look at the variability of
607 the Fe speciation, as was achieved by Ramkissoon *et al.* (2019) to create one or more modified
608 simulant compositions.

609 Near-IR spectroscopy indicated the simulant has many similarities to the CRISM data, with absorption
610 at 1.92 μm associated with the O-H stretch common to phyllosilicate minerals, in addition to
611 absorption at 2.39 μm (Clark *et al.* 1990). Absorption at 2.32 μm is associated with the Fe,Mg-OH
612 stretching and bending. However, the exact position depends on the Fe/Mg ratio and oxidation state
613 of Fe (Clark *et al.* 1990; Mustard 1992; Chemtob *et al.* 2015; Michalski *et al.* 2015). Comparing to the
614 Near-IR spectra for the components (Figure S1), the feature at 1.92 μm arises from the addition of
615 vermiculite, siderite and serpentine. While the feature at 2.32 μm arises from olivine, serpentine,
616 vermiculite and Fe-silicate.

617 5.1.3 Simulant Chemistry

618 Detailed geochemical analysis is difficult to obtain from orbit, therefore, this simulant is designed to
619 represent a possible mineralogy at Oxia Planum and not a geochemistry. While this simulant is a
620 mineralogical simulant and not a chemical simulant, the chemical composition was analysed using
621 SEM-EDS and XRF and found to be dominated by SiO_2 , MgO and FeO_T . Compared to the XRF data, the

622 calculated chemistry based on the SEM-EDS analysis of the components is as expected, with variations
623 of <8 % for all compounds; this variation is accepted given that the mineralogy was the target.

624 Taylor *et al.* (2010) and Gasnault *et al.* (2010) used Mars Odyssey's gamma ray spectrometer (GRS) to
625 identify geochemical provinces on Mars. Oxia Planum falls within a province extending from the
626 southeast of Acidalia Planitia to the northwest of Arabia Terra and is shown to have elevated Ca and
627 Fe content (Gasnault *et al.* 2010). Aside from the Al₂O₃ content, which is disproportionately high in the
628 simulant (as discussed above), the highest variation in oxide percentage between the simulant and
629 Sheepbed member is 3.32 wt. % (Table S5). The calculated CIA of the simulant was 51. A CIA above
630 40 indicates a weathered mineralogy. Values representative of an environment transitioning from a
631 closed-system to an open-system occur at 50-55 on Mars (Nesbitt and Young 1982; McLennan *et al.*
632 2014). The simulant CIA value is likely to be skewed upwards by its high Al₂O₃ content. For comparison,
633 it may be similar to the upper limit of the CIA values for Yellowknife Bay where CIA values between 35
634 and 45 correspond to closed system weathering (McLennan *et al.* 2014).

635 5.1.4 Simulant Physical Properties

636 The only constraint on the physical properties of this simulant was on the grain size used as it is
637 designed to represent mineralogy, however, these are important to understand as they may influence
638 the mechanical and thermal properties of future experiments. It should be noted that the simulant
639 production method of crushing, milling and sieving the simulant to a specific grain size, is likely to
640 have a large effect on its physical characteristics making then uncharacteristic of a natural sample.
641 From the grain size distribution, the simulant is classified as a clay-silt-sand, which is consistent with
642 the data from *Curiosity*; 53.1 % of the simulant falls within the silt-clay grain size, which is desirable
643 for a phyllosilicate -bearing terrain (Table 9). The remainder is a fine to medium sand.

644 The bulk density of the simulant was found to be 1.15 g cm⁻³. The bulk density of regolith at the
645 Pathfinder lander site was estimated to be between 2–2.2 g cm⁻³ (Moore *et al.* 1999) and drift material
646 at the Viking 1 landing site had an estimated bulk density of 1.15 ± 0.15 g cm⁻³ (Moore and Jakosky
647 1989), making the simulants bulk density on the lower side of what has been observed on Mars. This
648 is likely to be owing to the low density of the phyllosilicate minerals that dominate the simulant.
649 However, bulk density measurements for *in situ* phyllosilicate terrains on Mars have not yet been
650 made, and there have been no estimates for the bulk density of the material at Oxia Planum. The
651 porosity of the simulant was 61 %. The Viking missions' XRF suggested that porosity of the martian
652 regolith was between 31 and 58% (Moore and Jakosky 1989). This is likely due to the clay-silt-sand
653 particle distribution of the simulant.

654 5.2 Significance for rover exploration

655 5.2.1 Detectability of minerals at Oxia Planum

656 The development of this simulant highlights the potential difficulty in resolving individual mineral
657 phases from assemblages. The Raman analysis shown here is useful for informing the analysis by the
658 RLS instrument and is therefore relevant to the resolution of individual mineral phases. While
659 phyllosilicate minerals can be identified with Raman and Near-IR spectrometry, determining lower
660 weight percent minerals will be difficult without XRD. In addition, reactive Fe species give very broad
661 peaks or no peaks. Moreover, the small particles sizes meant obtaining clear peaks with Raman
662 spectroscopy was a challenge with a mixture comparable to a natural sample.

663 The Near-IR spectra for the simulant show some similarities to the CRISM and OMEGA data at Oxia
664 Planum as illustrated in Figure 4 (Carter *et al.* 2016; Mandon *et al.* 2021; Brossier *et al.* 2022; Turner
665 and Bridges 2017). These include absorption at 1.91 μm , associated with the presence of H_2O . While
666 the absorption at $\sim 2.3 \mu\text{m}$ is broader in the simulant than at Oxia Planum, suggesting the simulant Fe
667 content is higher than at Oxia Planum (Mustard *et al.* 2008). In addition, differences in spectral slope
668 can likely be accounted for by the powdered nature of the simulant as opposed to a consolidated rock
669 at Oxia Planum (Harloff and Arnold 2001). These overall similarities would suggest that the SOPHIA
670 simulant is a good match for Oxia Planum based on the spectral matches for the mineralogy from
671 orbit. However, despite the increased spectral resolution compared to CRISM, no new features are
672 resolved from the simulant spectra compared to the CRISM spectra, meaning additional instruments,
673 such as the RLS, may be needed to fully assess the mineralogy *in situ*.

674 5.2.2 Relevance to biosignature preservation at Oxia Planum

675 As the simulant contains a plausible mineral assemblage for Oxia Planum, it represents a possible
676 preservation environment for biosignatures. Consideration of the preservation potential of the
677 simulant is therefore important. The simulant contains 35 % vermiculite, which would likely be the
678 main preservation mineral at Oxia Planum. If the vermiculite is trioctahedral in nature it will likely have
679 a higher propensity to preserve organic matter due to its higher Cation Exchange Charge (CEC) when
680 compared to smectites (Jørgensen 1974; Krzesińska *et al.* 2021). The presence of Fe phases is also
681 strongly linked to the preservation of organic molecules, with, over 20% of organic carbon on Earth
682 being bound to reactive Fe mineral phases such as Fe (oxyhydr)oxides, which are closely associated
683 with phyllosilicates (Lalonde *et al.* 2012). Up to 60% of the materials investigated in Gale crater are
684 amorphous phases (Rampe *et al.* 2017, 2020). The Fe-rich endmember of these amorphous phases
685 may represent reactive Fe (Bonsall *et al.* 2022). The presence of iron rich amorphous material has
686 been shown to aid organic matter preservation (Lalonde *et al.* 2012), increasing the preservation
687 potential of the simulant.

688 In addition, the entire mineral assemblage of the simulant is relevant to the modification of biomarker
689 structures at Oxia Planum. Olivine and phyllosilicates have been shown to prevent biomarker
690 degradation during impacts and UV degradation respectively (Parnell *et al.* 2005; Bowden *et al.* 2009;
691 dos Santos *et al.* 2016; Montgomery *et al.* 2016; Furukawa *et al.* 2018). Both minerals have been
692 detected at Oxia Planum and are included in the simulant, the quantities of which will affect overall
693 biomarker preservation at Oxia Planum. This means experiments performed on the simulant will be
694 specific to the mineralogy. If the mineralogy at Oxia Planum is like the simulant design, it is likely to
695 have a high biomarker preservation potential.

696 5.2.3 Use in laboratory experiments

697 The SOPHIA simulant creates an opportunity for the community to perform a range of laboratory-
698 based experiments specific to Oxia Planum. As this simulant is designed to represent a plausible *in situ*
699 mineralogy at Oxia Planum, it is ideal for experimental studies investigating mineral interactions.
700 These include possible mineral-microbe interactions and habitability studies (e.g., Cockell *et al.* 2005;
701 Schuerger *et al.* 2012; Oliver *et al.* 2022). As the simulant represents the phyllosilicate terrains
702 preserved from Noachian-aged terrains at Oxia Planum, it can also be used in biosignature formation,
703 preservation, alteration and detectability studies specific to the payload on *Rosalind Franklin*. In
704 addition, in the absence of chemical data for Oxia Planum, this simulant can be used as a baseline for
705 the possible chemistry expected at Oxia Planum and used in geochemical investigations into rock-
706 water interactions at the site, further constraining its alteration environment (Krzyszewska *et al.* 2020;
707 Krzesińska *et al.* 2021).

708 Conclusion and Summary

709 We have designed a new martian mineralogical simulant for Oxia Planum, SOPHIA (Simulant for Oxia
710 Planum: Hydrated, Igneous, Amorphous), which is based on orbital data from Oxia Planum and
711 geologically comparable sites on Mars. SOPHIA possesses a hypothesised mineral assemblage for Oxia
712 Planum, which includes unaltered basaltic minerals, a phyllosilicate component and an amorphous
713 component. The characterisation of the simulant performed here using Near-IR shows that SOPHIA is
714 a good spectral match for Oxia Planum. This analysis, along with the Raman analysis, will assist in data
715 interpretation from the rover's RLS, ISEM and MicrOmega instruments. SOPHIA can be used in a range
716 of experiments concerning habitability and biomarker preservation, modification and detection in
717 anticipation of the mission. The simulant may be available for use upon request depending on
718 quantities required, the recipe and procedure described can also be modified for other experimental
719 purposes.

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